## SYNTHESIS OF TRICHOTOMINE, A BLUE PIGMENT OBTAINED FROM CLERODENDRON TRICHOTOMUM THUNB.

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Recently, the structure of trichotomine (I), a blue pigment obtained from <u>Clerodendron</u> <u>trichotomum</u> Thunb. was established in our laboratory.<sup>1</sup> We report here the synthesis of this blue pigment, trichotomine (I) which possesses the unique, new chromophore.

L-Tryptophan methyl ester<sup>2</sup> was reacted with succinic anhydride in anhydrous benzene (reflux, 5 hr), affording the amide (III),<sup>3,4</sup> mp 95 - 97°, (76%). The amide diester (IV),<sup>3,4</sup> mp 151 - 152° obtained on methylation of (III) ( $CH_2N_2$  - MeOH) was heated without solvent under reduced pressure (200°, 1 hr) to give the amorphous imide ester (V),<sup>3,5</sup> (90%). The cyclization of (V) was performed by employing phosphorus pentoxide in anhydrous benzene (reflux, 2 hr) to give the lactam ester (VI),<sup>3,6</sup> mp 208 - 211° (20% after recrystallization from EtOAc). The oxidative dimerization of (VI) was achieved by passing air gradually into



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the n-butanol solution of (VI) (90°, 18 hr), and the resulting mixture was separated by preparative thin layer chromatography [silica gel  $PF_{254}$ , EtOAc - benzene (1:2)] to afford the deep blue crystals (II),<sup>3</sup> mp 284 - 287° [35% based on reacted (VI)] together with the starting compound (VI) (~20% recovery). The identification of the synthetic compound (II) with natural trichotomine dimethyl ester (II),<sup>1a</sup> mp 285 - 287° was made by spectral (ir, uv, 100 MHz nmr, cd, and mass spectrum) and chromatographic comparison and by the mixture mp (284 - 287°).



The synthetic material (II) was hydrolyzed [1N KOH - MeOH - ether (1:1:1), room temperature, 3 hr] to give the amorphous compound, which was identical with the natural pigment,

trichotomine (I) by spectral and chromatographic comparison.

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## REFERENCES AND FOOTNOTES

- (a) S. Iwadare, Y. Shizuri, K. Sasaki, and Y. Hirata, <u>Tetrahedron Letters</u>, in the press;
  (b) K. Sasaki, S. Iwadare, and Y. Hirata, <u>Tetrahedron Letters</u>, in the press.
- 2. E. Abderhalden and M. Kempe, Hoppe-Seyler's Z. Physiol. Chem., 52, 207 (1907).
- 3. Elemental analysis or high resolution mass spectral data for this compound are in accord with theory.
- .4. This compound has been fully characterized by ir, nmr, and mass spectra confirmatory of the structure presented.
- 5. vmax (CHC1<sub>3</sub>) 3480, 1784, 1748, 1709 cm<sup>-1</sup>; δ (CDC1<sub>3</sub>, 100 MHz) 2.43 (4H, br.s), 3.62 (1H, dd, J = 16.5, 9.5 Hz), 3.61 (1H, dd, J = 16.5, 7.5 Hz), 3.80 (3H, s), 5.05 (1H, dd, J = 9.5, 7.5 Hz), 6.9 7.6 (4H, complex pattern, aromatic protons), 8.18 (1H, br.s); m/e 300 (M<sup>+</sup>).
- 6. vmax (KBr) 1747, 1683, 1654 cm<sup>-1</sup>;  $\lambda max$  (MeOH) nm ( $\varepsilon$ ) 231 (25,200), 308 (21,200), 321 (19,400);  $\delta$  (CDC13, 100 MHz) 3.18 (2H, d, J = 3.2 Hz), 3.27 (1H, dd, J = 16.0, 7.0 Hz), 3.65 (1H, dd, J = 16.0, 2.0 Hz), 3.67 (3H, s), 5.17 (1H, dd, J = 7.0, 2.0 Hz), 5.23 (1H, t, J = 3.2 Hz), 7.0 - 7.7 (4H, complex pattern, aromatic protons), 8.84 (1H, br.s); m/e 282 (M<sup>+</sup>); cd (MeOH) [ $\theta$ ]<sub>255</sub> +6340, [ $\theta$ ]<sub>320</sub> +2060, [ $\theta$ ]<sub>311</sub> -1510.